

**METHOD AND APPARATUS FOR PRODUCING  
NANO-PARTICLES OF SILVER**

Field of the Invention

5 This invention relates to the production of nano-  
particles in general and more particularly to a method and  
apparatus for producing nano-particles of silver (Ag).

Background

10 Nano-particles, that is, particles having average  
sizes less than about 1 micrometer (i.e., 1 micron) are  
known in the art and are of interest because their nano-  
crystalline and/or other nano-scale features dramatically  
change the properties of the material. For example,  
certain materials fabricated from nano-particles often  
possess superior mechanical properties compared with the  
15 same material fabricated in a conventional manner and with  
conventionally-sized starting materials (e.g., powders).  
Nano-particles of other materials may also possess unique  
electrical and/or magnetic properties, thereby opening the  
door to the fabrication of materials having previously  
20 unforeseen properties and attributes. The extremely large  
surface area to weight ratio of nano-particles allows  
nano-particles to interact with their surroundings very  
quickly which can also lead to the fabrication of new  
materials having new properties.

25 In sum, it is recognized that the ability to produce  
any material in nano-particle form represents a unique  
opportunity to design and develop a wide range of new and  
useful mechanical, optical, electrical, and chemical

applications, just to name a few. However, one problem that heretofore has limited the use of nano-particles is the difficulty in producing nano-particles of the desired size and composition on a commercial scale, e.g., by the kilogram instead of by the gram.

One method for producing nano-particles involves dissolving in a solvent precursor chemicals which define the composition of the final nano-particle product. The resulting composition is mixed to yield a solution which is substantially homogenous on a molecular level. The solvent is then evaporated at a sufficient rate so that the components in the homogenized solution are precipitated as a homogenized solid powder. While such wet processes have been used to produce nano-particles of various compositions, they are not without their problems. For example, such processes tend to produce larger particles along with the nano-particles, which must then be removed or separated from the nano-particles before the nano-particles can be used. Such wet processes can also involve a significant number of process steps and reagents which tend to increase the overall cost of the final nano-particle product.

Another method for producing nano-particles is a primarily mechanical process in which the precursor material is ground in a mill (e.g., a ball mill) until particles of the desired size are produced. Unfortunately, however, such grinding processes are energy

intensive, require substantial amounts of time, and typically result in the production of a powder containing not only the desired nano-particle product, but also particles having larger sizes as well. Of course, such  
5 larger sized particles must be separated from the nano-particles before they can be used. The abrasive materials used in such milling and grinding processes also tend to contaminate the nano-particle material. Consequently, such grinding processes generally are not conducive to the  
10 production of a highly pure nano-particle product.

Several other processes have been developed in which the precursor material is vaporized, typically in a partial vacuum, and then rapidly cooled in order to initiate nucleation and precipitate the nano-particle  
15 material. For example, in one process, a stream of vaporized precursor material is directed onto the surface of a cold (i.e., refrigerated) rotating cylinder. The vapor condenses on the cold surface of the cylinder. A scraper placed in contact with the rotating cylinder  
20 scrapes off the condensed material, which is then collected as the nano-particle product. In another process, the vapor stream of precursor material is condensed by expanding the vapor stream in a sonic nozzle. That is, the vapor stream is initially accelerated in the  
25 converging portion of the nozzle, ultimately reaching sonic velocity in the throat of the nozzle. The vapor stream is then further accelerated to a supersonic

velocity in the diverging section of the nozzle. The supersonic expansion of the vapor stream rapidly cools the vapor stream which results in the precipitation of nano-sized particles.

5           While the foregoing vaporization and cooling processes have been used to produce nano-particle materials, they are not without their problems. For example, the rotating cold cylinder process has proved difficult to implement on a large scale basis and has been  
10 less than successful in producing large quantities of nano-particle material. While the sonic nozzle process is theoretically capable of producing large quantities of nano-particles on a continuous basis, it requires the maintenance of a proper pressure differential across the  
15 sonic nozzle throughout the process. Another problem with the sonic nozzle process is that the nano-particle material tends to condense on the nozzle walls, which can seriously reduce the efficiency of the nozzle, and may even prevent it from functioning. While the condensation  
20 problem can be reduced by injecting a boundary layer stream along the nozzle walls, such a provision adds to the overall complexity and operational cost of the system.

Consequently, a need remains for a method and apparatus for producing nano-particles that does not  
25 suffer from the shortcomings of the prior art methods. Such a method and apparatus should be capable of producing large quantities of nano-particle product, preferably on

a continuous basis, and at a low cost. Ideally, such a method and apparatus should be less sensitive to certain process parameters than other systems, thereby allowing the method and apparatus to be more easily practiced on a large scale (i.e., commercial) basis. Additional advantages could be realized if the method and apparatus produced nano-particles in a relatively narrow size range, with a minimum amount of larger sized particles and/or contaminant materials.

#### Summary

Apparatus for producing silver nano-particle material according to one embodiment of the invention comprises a furnace and a crucible positioned within the furnace, the crucible containing a quantity of precursor material, the furnace heating the quantity of precursor material contained in the crucible to vaporize the precursor material. A process gas supply operatively associated with the furnace provides a process gas to an interior region of the furnace. A conduit is operatively associated with the furnace so that an inlet end of the conduit is open to the interior region of the furnace. A particle separator system is operatively associated with an outlet end of the conduit. A pump operatively associated with an outlet end of the particle separator system causes a mixture of process gas and vaporized precursor material contained in the interior region of the

furnace to be drawn into the inlet end of the conduit, the process gas cooling the vaporized precursor material to precipitate the silver nano-particle material in a carrier stream, the particle separator system separating the  
5 silver nano-particle material from the carrier stream.

Also disclosed is a method for producing silver nano-particle material comprising: Providing a precursor material in a crucible, the crucible being housed in a furnace; heating the precursor material in the crucible to  
10 produce a vaporized precursor material; contacting the vaporized precursor material with a process gas in a mixing region; drawing a mixture of vaporized precursor material and process gas into an inlet end of a conduit, the process gas cooling the vaporized precursor material  
15 to precipitate the silver nano-particle material in a carrier stream; and separating the silver nano-particle material from the carrier stream.

#### Brief Description of the Drawings

Illustrative embodiments of the invention are  
20 illustrated in the following drawings. The components in the drawings are not to scale, emphasis instead being placed upon clearly illustrating the principles of the present invention. Moreover, in the drawings, like reference numerals designate corresponding parts through  
25 the several views.

Figure 1 is a cross-sectional view in elevation of

the apparatus for producing silver nano-particle material according to one embodiment of the invention; and

Figure 2 is a transmission electron microscope image of silver nano-particle material produced by the present invention.

#### Detailed Description

Apparatus 10 for producing silver (Ag) nano-particle material 12 from a precursor material 14 is shown in Figure 1 and may comprise a furnace 16 defining an interior region 18 therein. A supply of process gas 20 is operatively associated with the furnace 16 and supplies process gas 22 to the interior region 18 of furnace 16. A crucible 24 positioned within the furnace 16 is provided with a quantity of the precursor material 14 therein. A conduit 26 having an inlet end 28 and an outlet end 30 is operatively associated with the furnace 16 so that the inlet end 28 of conduit 26 is open to the interior region 18 of the furnace 16. More specifically, the inlet end 28 of conduit 26 defines a hood-like portion 32 that is substantially vertically aligned over the crucible 24, as best seen in Figure 1. The area defined generally between the crucible 24 and the hood-like portion 32 of the inlet end 28 of conduit 26 comprises a mixing region 34.

A particle separator system 36 is operatively connected to the outlet end 30 of conduit 26 and is used to separate the silver nano-particle material 12 from a

particle-laden carrier stream 38 discharged from the outlet end 30 of conduit 26. More specifically, the particle separator system 36 comprises an inlet end 40, an outlet end 42, and a material discharge port 44. The inlet end 40 of particle separator system 36 is connected to the outlet end 30 of conduit 26. The outlet end 42 of particle separator system 36 is connected to a vacuum pump 46. The material discharge port 44 is connected to a product collection chamber 48. In one embodiment, the particle separator system 36 is also provided with a filter assembly 50 to filter or remove the silver nano-particle material 12 from the particle-laden carrier stream 38 and prevent the silver nano-particle material 12 from entering the vacuum pump 46. Consequently, substantially all of the silver nano-particle material 12 produced by the apparatus 10 will be available for recovery.

In the embodiment shown and described herein, the precursor material 14 contained in the crucible 24 is heated to the point of vaporization by an induction coil 52 surrounding the crucible 24. The induction coil 52 is electrically connected to a power supply 54. As will be described in greater detail below, the power supply 54 provides a high-frequency alternating current to the induction coil 52, which results in the heating and ultimate vaporization of the precursor material 14 via a process known as induction heating.



The apparatus 10 for producing silver nano-particle material 12 may be operated as follows. A supply of precursor material 14, for example, silver metal, is provided to the crucible 24. In the embodiment shown and described herein, the precursor material 14 may be supplied on a continuous basis by a precursor material feed system 60 in the manner that will be described below. Alternatively, the precursor material 14 may be supplied in batches. The vacuum pump 46 may be activated and a flow of process gas 22 from the process gas supply 20 initiated. The flow of process gas 22 may be adjusted in accordance with the teachings provided herein to allow the interior 18 of the furnace 16 to reach a pressure within the range of operating pressures specified herein. Exemplary pressures include pressures within the range of about 0.1 to about 6 torr. The power supply 54 may then be activated to begin heating the precursor material 14 to cause the same to vaporize. Exemplary heating temperatures include temperatures in the range of about 1600 to about 2000 °C. Vaporized precursor material 56 then drifts generally upward toward the hood 32 defined by the inlet end 28 of conduit 26, whereupon it contacts the process gas 22, generally within the mixing region 34. The resulting mixture 58 of process gas 22 and vaporized precursor material 56 is drawn into the inlet end 28 of conduit 26 by the action of vacuum pump 46. As the process gas 22 combines with the vaporized precursor

material 56, the process gas 22 begins to cool the vaporized precursor material 56, resulting in the precipitation of the silver nano-particle material 12. In this regard it should be noted that the process of cooling  
5 generally begins in the mixing region 34 and continues as the mixture 58 flows through the conduit 26 so that substantially all available vaporized precursor material 56 contained in the mixture 58 precipitates as the silver nano-particle material 12. Thereafter, the precipitated  
10 silver nano-particle material 12 is suspended in the particle-laden carrier stream 38 which is discharged by the outlet end 30 of conduit 26. The silver nano-particle material 12 is separated from the carrier stream 38 by the particle separator system 36 and collected in the product  
15 collection chamber 48.

The silver nano-particle material 12 produced according to the method and apparatus of the present invention may be imaged in accordance with any of a wide range of microscopy processes that are now known in the  
20 art or that may be developed in the future that are suitable for imaging particles in the nano-size range. For example, Figure 2 is an image of the silver nano-particle material 12 produced by a transmission electron microscope in a process generically referred to as  
25 transmission electron microscopy (TEM). As is readily seen in the TEM image illustrated in Figure 2, each individual particle of the silver nano-particle material

12 comprises a generally spherical configuration. While the size of the silver nano-particle material 12 can be expressed in terms of the mean diameter of the particles (e.g., as imaged by transmission electron microscopy), it is generally more useful to express the size of the silver nano-particle material 12 in terms of surface area per unit weight. Measurements of particle surface area per unit weight may be obtained by BET analysis. As is well-known, BET analysis involves an extension of the Langmuir isotherm equation using multi-molecular layer absorption developed by Brunauer, Emmett, and Teller. BET analysis is an established analytical technique that provides highly accurate and definitive results. In the embodiment shown and described herein, the method and apparatus of the present invention has produced silver nano-particle material 12 having sizes in the range of about 10 to about 20 square meters/gram ( $\text{m}^2/\text{g}$ ), or about 57 to about 28 nanometers. Alternatively, other types of measuring processes may be used to determine the particle size.

A significant advantage of the present invention is that it can be used to produce the silver nano-particle material 12 in very large quantities and at a very low cost. The present invention is also relatively simple to construct, easy to operate, and is not overly sensitive to process parameters. In addition, the ability to supply the precursor material 14 on a continuous basis allows the present invention to be readily adapted for use in large-

scale (e.g., commercial) production. The silver nano-particle material 12 produced by the present invention also contains particles within a fairly narrowly-defined size range and with a minimum amount of larger-sized particles. Consequently, the silver nano-particle material 12 produced in accordance with the method and apparatus of the present invention may be generally regarded as a high quality product that requires little or no additional processing before it may be used.

Another advantage of the present invention is that it is generally immune to problems associated with the condensation of the silver nano-particle material 12 on the internal surfaces of conduit 26. Consequently, it is not necessary to take additional measures to discourage the accumulation of the silver nano-particle material 12 on the internal surfaces of the conduit 26. However, even if the silver nano-particle material 12 eventually accumulates on the internal components of the system, the simple design of the system will allow any such accumulation to be easily removed.

Having briefly described the method and apparatus according to one embodiment of the present invention, as well as some of their more significant features and advantages, the various embodiments of the method and apparatus for producing nano-particles of the present invention will now be described in detail.

Referring back now to Figure 1, one embodiment of the

apparatus 10 for producing silver nano-particle material 12 may comprise a furnace 16 having an interior region 18 defined therein. In the embodiment shown and described herein, the furnace 16 comprises a generally rectangular configuration, although other shapes or configurations are possible. The furnace may be constructed from any of a wide range of materials suitable for the intended application. Consequently, the present invention should not be regarded as limited to furnaces constructed of any particular material. However, by way of example, in one preferred embodiment, the furnace 16 is constructed from mild steel lined with copper. The interior walls of the furnace 16 may also be lined with a refractory material (e.g., mica, fire brick, etc.) (not shown) to insulate the furnace and to prevent the furnace walls from being adversely affected by the extreme heat contained therein.

The furnace 16 is also adapted to receive a flow of process gas 22. In the embodiment shown and described herein, the process gas 22 may be supplied to the interior region 18 of furnace 16 via a tube or pipe 62 connected to a process gas supply 20. The flow of process gas 22 from the process gas supply 20 may be controlled by a suitable valve/pressure regulator assembly 64 of the type well-known in the art and readily commercially available. In the embodiment shown and described herein, the combined action of the vacuum pump 46 and flow of process gas 22 should be such that the pressure of the interior region 18

of the furnace 16 can be maintained at a pressure in the range of about 0.1 to about 6 torr (3 torr preferred) during the production process. The process gas 22 may comprise any of a wide range of gases, such as, for  
5 example, nitrogen, argon, or combinations thereof. Alternatively, other inert gases, such as helium, neon, krypton, or xenon may also be used. By way of example, in one preferred embodiment the process gas 22 comprises N<sub>2</sub>. Accordingly, the process gas supply 20 may comprise a  
10 supply of nitrogen, either in liquid or gas form. However, because apparatus for supplying such process gases are well-known in the art and could be readily provided by persons having ordinary skill in the art, the supply of process gas 20 and associated components will  
15 not be described in further detail herein.

The crucible 24 provided within the furnace 16 receives the precursor material 14 and serves to contain the precursor material 14 while it is heated and, ultimately, vaporized within the crucible 24. In the  
20 embodiment shown and described herein, the crucible 24 comprises a generally cup-shaped member having an open top through which to receive the precursor material 14 and through which escapes the vaporized precursor material 56. The crucible 24 may be supported within the furnace 16 by  
25 any convenient means, such as a support tube 68. The crucible 24 and support tube 68 should be made from materials suitable for withstanding the temperatures

required to vaporize the precursor material 14 (e.g., temperatures in the range of about 1600°C to about 2000°C). By way of example, in one preferred embodiment, the crucible 24 comprises carbon graphite, whereas the support tube 68 comprises quartz, although other materials could be used. The carbon graphite material used in one preferred embodiment comprises No. ISO-63, commercially available from Advanced Graphite Engineering of Greenville, Michigan 48838. If the crucible 24 comprises carbon graphite or other reactive material (that is not desired to react with the precursor material 14), the interior surfaces of crucible 24 should be provided with a non-reactive coating or liner 66 to prevent the graphite or other reactive material from reacting with the precursor material 14. By way of example, in one preferred embodiment, the non-reactive coating or liner 66 comprises ceramic item No. AC6043, commercially available from GE Advanced Ceramics of Strongsville, Ohio 44149. Alternatively, such a non-reactive liner 66 would not be required if the entire crucible 24 were to be fabricated from a non-reactive material.

The crucible 24 may comprise any of a wide range of sizes depending on the capacity and desired production rate of the apparatus 10. Consequently, the present invention should not be regarded as limited to a crucible 24 having any particular size. However, by way of example, in one preferred embodiment, the crucible 24 has

an outside diameter of about 8 cm and a height of about 8 cm.

As mentioned above, in one embodiment, the precursor material 14 provided in the crucible 24 is heated via induction heating. Accordingly, the furnace 16 is provided with an induction coil 52 therein that surrounds the crucible 24. In this arrangement, the crucible 24 functions as the susceptor for the induction coil 52 that surrounds the carbon graphite crucible 24. In the embodiment shown and described herein, the induction coil 52 comprises several turns (e.g., 4 turns) of copper tubing through which water is caused to flow as a coolant. Each end of the coiled copper tubing is connected to a power supply 54 which supplies alternating current to the copper tubing comprising the induction coil 52. In one embodiment, the power supply 54 supplies alternating current power to the induction coil 52 at voltage of about 105 volts, a current of about 9 kilowatts, and a frequency of about 10,000 Hz. However, because induction coils and power supplies for providing electrical power to such induction coils are well-known in the art and could be readily supplied by persons having ordinary skill in the art after having become familiar with the details of the present invention, the particular induction coil and power supply that may be utilized in embodiments of the invention will not be described in further detail herein.

In the embodiment shown and described herein, an



insulating material 78 is provided between the induction coil 52 and crucible 24. The insulating material 78 prevents excessive heat from being transferred between the crucible 24 to the induction coil 52. The insulating material 78 may comprise any of a wide range of materials suitable for the high temperatures involved. By way of example, in one preferred embodiment, the insulating material comprises a carbon-black felt material.

The apparatus 10 may also be provided with a infrared temperature sensor 80 suitable for sensing the temperature of the precursor material 14 contained within the crucible 24. Alternatively, other types of temperature sensors may also be used to allow the temperature of the crucible 24 and/or precursor material A contained therein to be monitored.

The apparatus 10 may also be provided with a precursor material feed system 60 to provide a continuous supply of precursor material 14 to the crucible 24. In one embodiment, the precursor material feed system 60 comprises a supply (e.g., a spool) 70 of wire 72. The feed system 60 may also be provided with a wire-feed speed regulator system (not shown) to regulate or control the speed at which the wire 72 is fed into the crucible 24. However, because systems for feeding wire and systems for feeding other forms of precursor material 14 (e.g., powder, pellets, etc.) are known in the art and could be easily provided by persons having ordinary skill in the

art after having become familiar with the teachings provided herein, the particular components and devices that may be utilized in the precursor material feed system 60 will not be described in further detail herein.

5           The conduit 26 comprises a generally cylindrically-shaped tube-like member having an inlet end 28 and an outlet end 30. The conduit 26 is operatively associated with the furnace 16 so that the inlet end 28 is positioned within the interior region 18 defined by the furnace 16  
10           and so that the outlet end 30 is exterior to the furnace 16. A portion of the inlet end 28 of conduit 26 is cut-away to form a hood-like portion 32 that is substantially vertically aligned over the crucible 24. Alternatively, a separate member could be provided that defines the hood-  
15           like portion 32 of inlet end 28. As mentioned above, the area defined generally between the crucible 24 and the hood-like portion 32 of the inlet end 28 of conduit 26 comprises mixing region 34.

          The conduit 26 may have dimensions that are  
20           commensurate with the size (i.e., desired production capacity) of the apparatus 10 for producing silver nanoparticle material 12. Accordingly, the present invention should not be regarded as limited to a conduit 26 having any particular size. However, by way of example, in one  
25           preferred embodiment, the conduit 26 has an inside diameter ranging from about 90 mm inside the furnace 16 to a diameter of about 51 mm outside the furnace 16, and a

total length of about 865 mm. The conduit 26 may be fabricated from any of a wide range of materials suitable for the intended application, keeping in mind the extreme temperatures (e.g., in the range of about 1600°C to about 2000°C) existing within the interior region 18 of the furnace 16. Accordingly, in one preferred embodiment, the conduit 26 is fabricated from quartz. However, it should be noted that the conduit 26 need not comprise the same material throughout, but instead could comprise different materials depending on whether they are located within without the furnace 16. For example, in the embodiment shown and described herein, the portion of the conduit 26 contained within the furnace 16 comprises quartz. However, the portion of the conduit 26 located outside of the furnace 16 may be fabricated from other materials since the outside portion of the conduit 26 will not be exposed to extreme temperatures. For example, in one preferred embodiment, the portion of the conduit 26 exterior to the furnace 16 is fabricated from stainless steel. The two different sections (e.g., the quartz section and the stainless steel section) of the conduit 26 may be joined together by any convenient means, as would be obvious to persons having ordinary skill in the art after having become familiar with the teachings of the present invention.

The particle separator system 36 is operatively associated with the conduit 26 and serves to separate the

silver nano-particle material 12 from the carrier stream 38 discharged by the conduit 26. In one embodiment, the particle separator system 36 comprises a generally cylindrically-shaped housing 74 having an inlet end 40, an outlet end 42, and a material discharge port 44. The housing 74 is also sized to receive a filter 50. The inlet end 40 of housing 74 is connected to the outlet end 30 of conduit 26 in the manner illustrated in Figure 1. The outlet end 42 of housing 74 is fluidically connected to vacuum pump 46. The material discharge port 44 is located at the lower portion of the housing 74 and may be connected to a product collection chamber 48.

The filter 50 contained in the housing 74 may comprise any of a wide range of filter materials known in the art and readily commercially available that would be effective for separating or filtering the silver nano-particle material 12 from the carrier stream 38 to substantially prevent the silver nano-particle material 12 from entering the vacuum pump 46. Consequently, the present invention should not be regarded as limited to a filter 50 comprising any particular material. However, by way of example, in one preferred embodiment, the filter element 50 comprises a filter element available from the Grainger Company as filter canister no. 6JD03, and filter cartridge no. 6JC96. The filter 50 may be harvested from time-to-time to remove accumulated silver nano-particle material 12.

In an alternative embodiment, the particle separator system 36 may comprise a liquid scrubber-type filter (not shown) wherein the silver nano-particle material 12 is collected by bubbling the carrier stream 38 through a liquid. This type of particle separator system could be used to advantage if a slurry containing the silver nano-particle material is the desired product. Any of a wide range of liquids could be used, again depending on the nature of the slurry product desired. Alternatively, the liquid and silver nano-particle material could be separated by conventional techniques if so desired. Still other filtering devices and other separation processes (whether with or without filters) are possible and could be used to capture and remove the silver nano-particle material 12 from the carrier stream 38, as would be obvious to persons having ordinary skill in the art after having become familiar with the teachings of the present invention. Consequently, the present invention should not be regarded as limited to the particular product collection apparatus shown and described herein.

The vacuum pump 46 may comprise any of a wide range of vacuum pumps known in the art and readily commercially available that would be suitable for maintaining the pressure within the furnace within the ranges specified herein. By way of example, in one preferred embodiment, the vacuum pump 46 comprises a SDS 800 type vacuum pump, commercially available from Stokes Vacuum, Inc. of

Philadelphia, Pennsylvania 19120.

The apparatus 10 may be operated in accordance with the following method in order to produce silver nanoparticle material 12. As mentioned, the silver nanoparticle material 12 is produced by vaporizing a precursor material 14 which may comprise metallic silver. In the embodiment shown and described herein, the metallic silver precursor material 14 is derived from silver wire 72 that is fed into the crucible 24 on a continuous basis by the precursor material feed system 60. Alternatively, the precursor material 14 may be provided in batches. Once the precursor material 14 is delivered to the crucible 24 contained within the furnace 16, the induction coil 52 heats the precursor material 14 to a temperature in the range of about 1600°C to about 2000°C, which is sufficient to vaporize the silver precursor material 14, resulting in the production of a vaporized precursor material 56. Optimum results may be obtained within a temperature in the range of about 1700 °C to about 1800 °C. The pressure within the furnace is maintained by adjusting the flow of process gas 22 and/or the vacuum pump 46 to maintain the internal pressure of the furnace 16 within a range of about 0.1 torr to about 6 torr, with optimum results being obtained with a pressure in the range of about 2 torr to about 4 torr.

The vaporized precursor material 56 drifts generally upwardly out of the crucible 24 and enters the mixing

region 34, whereupon it begins to mix with the process gas 22. By way of example, in one preferred embodiment, the process gas 22 may comprise nitrogen. However, other process gases, such as any of the inert gases (e.g., helium, neon, argon, krypton, xenon, or mixtures thereof), may also be used. Alternatively, mixtures of nitrogen and any of the inert gases may also be used. The mixing of the vaporized precursor material 56 and the process gas 22 results in the formation of the mixture 58. In addition, the process gas 22 begins to cool the vaporized precursor material 56, thus beginning the precipitation within the mixture 58 of the silver nano-particle material 12. The mixture 58 of process gas 22, vaporized precursor material 56, and amounts of precipitated silver nano-particle material 12 enters the inlet end 28 of conduit 26 by the action of vacuum pump 46. Additional amounts of the silver nano-particle material 12 continue to precipitate from the mixture 58 until, by reaching the outlet end 30 of conduit 26 substantially all of the vaporized precursor material 56 has been precipitated as the silver nano-particle material 12. The precipitated silver nano-particle material 12 exits the outlet end 30 of conduit 26 as particle-laden carrier stream 38. Thereafter, the silver nano-particle material 12 is ultimately separated from the particle-laden carrier stream 38 by the particle separator system 36. The remaining carrier stream 38 passes through the filter 50 and vacuum pump 46 and is

discharged into the surrounding atmosphere as filtered carrier stream 76. The silver nano-particle material 12 is collected within the lower portion of the housing 74 of particle separator system 36 and is ultimately discharged through the material discharge port 44 into the product collection chamber 48. The filter 50 may be harvested from time-to-time to collect additional amounts of silver nano-particle material 12 that may have accumulated on the filter 50.

Figure 2 is an image of the silver nano-particle material 12 produced by a transmission electron microscope in a process that is commonly referred to as transmission electron microscopy (TEM). As is readily seen in Figure 2, each individual particle of the nano-particle material 12 comprises a generally spherically-shaped configuration. While the size of the nano-particle material 12 can be expressed in terms of the mean diameter of the particles (e.g., as detected by transmission electron microscopy), it is generally more useful to express the size of the silver nano-particle material 12 in terms of surface area per unit weight. Measurements of particle surface area are per unit weight may be obtained by BET analysis which is, as mentioned above, an established analytical technique that provides highly accurate and definitive results. In the embodiment shown and described herein, the method and apparatus of the present invention has been used to produce a nano-particle material having a size in



the range of about 10 to about 20 square meters/gram ( $\text{m}^2/\text{g}$ ), as measured in accordance with the BET analysis technique, or about 57 to about 28 nanometers. Alternatively, other types of measuring processes may be  
5 used to determine the particle size.

#### EXAMPLE

In this Example, the apparatus 10 comprised the various components and having the configurations described herein. The precursor material 14 comprised metallic  
10 silver wire 72 having a diameter of 3.18 mm (0.125 inches) and a purity of about 99.9%. The process gas 22 comprised nitrogen and the flow thereof was regulated to provide a pressure within the furnace 16 of about 3 torr. The silver wire 72 was fed into the crucible 24 in a  
15 continuous manner at a rate of about 360 g/hour. Once within the crucible, the precursor material 14 was heated to a temperature of about  $1725^\circ\text{C}$  which was sufficient to vaporize the silver precursor material 14. Once the vaporized precursor material 56 began to contact the  
20 process gas 22, the apparatus 10 started to produce the silver nano-particle material 12. The silver nano-particle material 12 was then separated by the separator 36 as described hereinabove and ultimately filtered or separated from the carrier stream 38. The silver nano-  
25 particle material 12 was collected in the product collection chamber 48 as described above. The apparatus

10 was operated in this manner for a time period of about 60 minutes, which resulted in the production of about 270 g of silver nano-particle material 12.

5        Having herein set forth preferred embodiments of the present invention, it is anticipated that suitable modifications can be made thereto which will nevertheless remain within the scope of the invention.